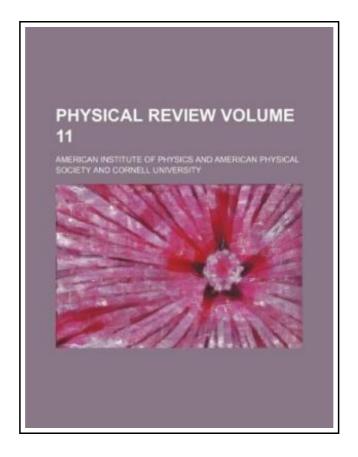
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Most of these publication is the greatest publication offered. It is actually rally intriguing throgh reading period of time. You can expect to like just how the article writer create this publication. (Eddie Schuppe)

PHYSICAL REVIEW VOLUME 11



Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1900 Excerpt: .(Fig. 17), and is a straight line as other observers have shown it to be. It will be noticed that the curve continues a straight line on through the boiling point and for some distance above. It was pleasing to have the two curves gotten under such different conditions join so perfectly. This seems to indicate conclusively that the apparatus was capable of giving accurate results, and that the curve above 100 degrees undoubtedly represents the true values of the surface tension. The current required to lift the system in air at atmospheric pressure was 96.4 mv., at 2000 C, the current required to lift the system in the vapor was 96 mv., at 3000 C, 95 mv., and at 3600 C, 91 mv. From these values, which are taken from the base line, it is evident that the curve a is displaced by an amount equal to the drop in the base line. In the curve c (Fig. 17) temperatures are platted as abscissas and differences (between the curves a and b) as ordinates. This gives the true form of the curve, the data for which is contained in table VII. 1 Landolt and Bornstein, Phys.-Chem. Tables, p. 44. The surface tension can be expressed in any system of units when once its value is known for any given temperature, i. e., if we accept 74.22 dynes as its value at 200 C., then the surface tension at, say, 200 C. is 19.3 x 1.78...



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